# Mar. 1978 Reaction of Ketenes with N,N-Disubstituted $\alpha$ -Aminomethyleneketones VII. Synthesis of 2H,5H-[1]Benzothiopyrano[4,3-b] pyran and 2H,5H-Thiopyrano[4,3-b] pyran Derivatives

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The dipolar 1,4-cycloaddition of dichloroketene to N,N-disubstituted 3-aminomethylene-2,3-dihydro-4-thiochromanones and 3-aminomethylenetetrahydro-4-thiopyranones gave N,N-disubstituted 4-amino-3,3-dichloro-3,4-dihydro-2H,5H-[1]benzothiopyrano[4,3-b]pyran-2-ones and 4-amino-3,3-dichloro-3,4,7,8-tetrahydro-2H,5H-thiopyrano[4,3-b]pyran-2-ones, respectively, only in the case of aromatic or strong hindering aliphatic N-substitution. The adducts gave N,N-disubstituted 4-amino-3-chloro-2H,5H-thiopyrano[4,3-b]pyran-2-ones and 4-amino-3-chloro-7,8-dihydro-2H,5H-thiopyrano[4,3-b]pyran-2-ones, respectively, by dehydro-chlorination with DBN. By chromatography on neutral alumina, 3-(2,2-dichloroethylidene)-2,3-dihydro-4-thiochromanone was isolated as an unstable liquid from the reaction between dichloroketene and 3-diethylaminomethylene-2,3-dihydro-4-thiochromanone.

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In a previous publication we described the reaction of N, N-disubstituted 3-aminomethylene-2,3-dihydro-4-thio-chromanones I and 3-aminomethylene-tetrahydro-4-thio-pyranones IV with sulfene to give new polycondensed sulfur heterocycles (1).

As part of our continuing study of the cycloadditions of N,N-disubstituted  $\alpha$ -aminomethyleneketones to dichloroketene (2-6), we wish to report the synthesis of 2H,5H-[1]benzothiopyrano[4,3-b]pyran and 2H,5H-thiopyrano [4,3-b] pyran derivatives starting from I and IV. For this purpose, we reacted enamines Ib,e,f,IVc (1), IVd,e (the last two are new compounds, see Experimental) with dichloroacetyl chloride and triethylamine (dichloroketene prepared in situ). As in similar cases, ie., aromatic or strong hindering aliphatic N-substitution of the αaminomethyleneketone (2-6), the dipolar 1,4-cycloaddition occurred readily to give N,N-disubstituted 4-amino-3, 3-dichloro-3, 4-dihydro-2H, 5H-[1]benzothiopyrano-[4,3-b]pyran-2-ones IIa-c and 4-amino-3,3-dichloro-3,4,7,8-tetrahydro-2H,5H-thiopyrano[4,3-b]pyran-2-ones Va-c. The structures of II and V were supported by uv, ir, and nmr spectral data (Tables I and III).

Refluxing IIa-c and Va-c with 1,5-diazabicyclo [4.3.0]-non-5-ene (DBN) in benzene (4) gave the dehydrochlorinated products, namely N,N-disubstituted 4-amino-3-chloro-2H,5H-[1]benzothiopyrano [4,3-b]pyran-2-ones IIIa-c and 4-amino-3-chloro-7,8-dihydro-2H,5H-thiopyrano [4,3-b]-pyran-2-ones VIa-c (Tables II and IV).

The reaction of I and IV bearing aliphatic N,N-disubstitution with dichloroketene was tried only in the case of Ia and IVa  $[NR_2 = N(C_2H_5)_2]$ . Whereas IVa led to inconclusive results, Ia apparently gave the same reaction as in another case previously described by us (5). By chromatography on neutral alumina, an unstable liquid was obtained in a low yield, which was identified

as 3-(2,2-dichloroethylidene)-2,3-dihydro-4-thiochromanone VII by its ir and nmr spectral data.

In conclusion, as a survey of the literature revealed, few derivatives of 2H,5H-[1] benzothiopyrano [4,3-b]-pyran (7,8) and especially of 2H,5H-thiopyrano [4,3-b]-pyran (9) are known. This new example of dipolar cycloaddition of heterocyclic  $\alpha$ -aminomethyleneketones to dichloroketene seems to open another route to the synthesis of the above-mentioned sulfur heterocycles.

### **EXPERIMENTAL**

Uv spectra were measured in 95% ethanol with a Hitachi-Perkin-Elmer Model EPS-3T spectrophotometer. Ir spectra

Table I

N,N-Disubstituted 4-Amino-3,3-dichloro-3,4-dihydro-2H,5H-[1] benzothiopyrano[4,3-b] pyran-2-ones Ha-c

					Analyses % Calcd./Found			
Compound No.	$NR_2$	Yield %	M.p., °C	Molecular Formula	C	Н	N	
IIa	$N[CH(CH_3)_2]_2$	83	151 (a)	$C_{18}H_{21}Cl_2NO_2S$	55.96 56.05	5.48 5.60	3.62 3.60	
Hb	$N(CH_3)C_6H_5$	81	149 (a)	$C_{19}H_{15}Cl_{2}NO_{2}S$	58.17 58.02	3.85 4.04	$\frac{3.57}{3.40}$	
Пе	$N(C_6H_5)_2$	86	177 (a)	$C_{24}H_{17}Cl_2NO_2S$	63.44 63.45	3.77 3.93	3.08 2.93	
		Uv, Ir	and Nmr Spectr	al Data				
	Uv $\lambda$ max nm (log $\epsilon$ )  Ir, cm <sup>-1</sup> C=O C=C		Νπ, δ					
Ha	236 (4.25), 255 sh (4.1 340 (3.38)	13), 178	7 1667 (b)	1.15 (d, $J = 6.6$ , 4CH <sub>3</sub> ), 3.11 (h, $J = 6.6$ , 2NCH), 3.59 and 3.79 (2d, $J = 15$ , CH <sub>2</sub> ·5), 3.86 (near s, H-4), 7.28 (m, 3H <sub>ar</sub> ), 7.65 (m, H <sub>ar</sub> ·10) (d)				
IIb	242 (4.48), 350 (3.22)	178	88 1678 (c)	2.80 (s, NCH <sub>3</sub> ), 3.36 and 3.67 (2d, J = 15, CH <sub>2</sub> -5), 5.00 (near s, H-4), 7.34 (m, 8H <sub>ar</sub> ), 7.72 (m, H <sub>ar</sub> -10) (d)				
He	240 (4.45), 262 sh (4.5	27), 179	4 1680 (b)	3.54 and 3.91 (2d, $J = 14.4$ , $CH_2-5$ ), 5.20				

(a) From ethyl acetate. (b) In tetrachloromethane. (c) In potassium bromide. (d) In deuteriochloroform.

were taken on a Perkin-Elmer Model 257 spectrophotometer; nmr spectra were recorded on a Perkin-Elmer Model R12 instrument. Chemical shifts are reported as  $\delta$  (ppm) relative to TMS as an internal standard; J in Hz. Melting points were determined with a Mettler FP1 apparatus.

290 sh (4.06), 340 (3.42)

Compounds Ila-c and Va-c were prepared according to (2) and were dehydro chlorinated with DBN according to (4).

3- Diisopropylaminomethylenetetrahydro-4-thiopyranone (IVd) and 3-Diphenylaminomethylenetetrahydro-4-thiopyranone (IVe).

These compounds were synthesized by a previously described procedure (10).

## Compound IVd.

This compound was obtained in a yield of 65%, m.p.  $100^{\circ}$  from anhydrous diethyl ether; uv  $\lambda$  max nm (log  $\epsilon$ ): 334 (4.29); ir (tetrachloromethane)  $\nu$  max: 1642, 1525 cm<sup>-1</sup>; nmr (tetrachloromethane):  $\delta$  1.28 (d, J = 6.6, 4CH<sub>3</sub>), 2.65 (m, CH<sub>2</sub>-5 + CH<sub>2</sub>-6), 3.59 (near s, CH<sub>2</sub>-2), 3.86 (h, J = 6.6, 2CHN), 7.50 (near s, =CHN).

Anal. Calcd. for C<sub>12</sub>H<sub>21</sub>NOS: C, 63.39; H, 9.31; N, 6.16. Found: C, 63.20; H, 9.38; N, 6.05.

## Compound IVe.

This compound was obtained in a yield of 78%, m.p. 94°

from anhydrous diethyl ether; uv  $\lambda$  max nm (log  $\epsilon$ ): 232 (3.99), 281.5 (3.99), 354 (4.30): ir (tetrachloromethane)  $\nu$  max: 1660, 1545 cm<sup>-1</sup>; nmr (tetrachloromethane):  $\delta$  2.78 (m, 3CH<sub>2</sub>), 7.19 (mc, 2NC<sub>6</sub>H<sub>5</sub>), 7.57 (near s, =CHN).

(near s, H-4), 7.15 (mc, 14H<sub>ar</sub>) (b)

Anal. Calcd. for  $C_{18}H_{17}NOS$ : C, 73.19; H, 5.80; N, 4.74. Found: C, 73.32; H, 6.08; N, 5.02.

This enamine is an E-isomer, as can be seen from the strong upfield shift of  $CH_2$ -2 protons ( $\sim$ 0.8 ppm) caused by the phenyl groups [compare (1)].

## 3-(2,2-Dichloroethylidene)-2,3-dihydro-4-thiochromanone (VII).

This compound was obtained starting from Ia (1), dichloroacetyl chloride and triethylamine in anhydrous benzene, by a previously described procedure (5). Chromatography on neutral alumina grade I followed by elution with petroleum ether (b.p. 40-70°) gave VII (36%) as an unstable liquid; further elution with benzene and diethyl ether gave no results. Because VII decomposed by attempted distillation in vacuo and even by standing at 0°, becoming deep-green coloured, it was purified by chromatography; ir (tetrachloromethane)  $\nu$  max: 1627, 1674 cm<sup>-1</sup> [cf. (5)]; nmr (tetrachloromethane):  $\delta$  3.88 (near s, CH<sub>2</sub>-2), 6.56 and 6.89 (2d, J = 9.6, =CH-CHCl<sub>2</sub>), 7.30 (m, 3H<sub>ar</sub>), 7.95-8.30 (m, H<sub>ar</sub>-5) [cf. (5)].

Anal. Calcd. for  $C_{11}H_8Cl_2OS$ : C, 50.98; H, 3.11. Found: C, 51.06; H, 3.19.

 $\label{limino-3-chloro-2H,5H-[1]} \mbox{$T$able II} $$N,N$-Disubstituted $4$-Amino-3-chloro-$2H,5H-[1]$ benzothiopyrano[4,3-$b]$ pyran-2-ones IIIa-c$ 

					Analyses % Calcd./Found		
Compound No.	$NR_2$	Yield %	M.p., °C	Molecular Formula	С	H	N
IIIa	$N[CH(CH_3)_2]_2$	81	122 (a)	$\mathrm{C_{18}H_{20}CINO_{2}S}$	$61.79 \\ 62.11$	5.76 5.61	$\frac{4.00}{4.27}$
Шь	$N(CH_3)C_6H_5$	96	142 (a)	$C_{19}H_{14}CINO_2S$	$64.13 \\ 64.17$	$\begin{array}{c} 3.97 \\ 4.05 \end{array}$	3.94 3.80
IIIc	$N(C_6II_5)_2$	74	190 (a)	$C_{24}H_{16}CINO_2S$	68.98 68.94	3.86 3.90	3.35 3.30

## Uv, Ir and Nmr Spectral Data

	Uv $\lambda$ max nm (log $\epsilon$ )	Ir, em <sup>-1</sup> (b) C=O C=C	Nmr, δ
IIIa	241.5 (4.36), 270 (3.60), 340 (3.99), 380 (3.97)	1738 1618 1510	1.22 (2 superimposed d, J = 6.6, 4CH <sub>3</sub> ), 3.77 (near h, J = 6.6, 2NCH), 3.87 (near s, CH <sub>2</sub> -5), 7.25 (m, $3H_{ar}$ ), 7.85 (m, $H_{ar}$ -10) (b)
ШЬ	243 (4.51), 274 (3.87), 301 sh (3.70), 344 (4.05), 387 (4.11)	1738 1623 1515	3.42 (near s, $CH_2$ -5 + $NCH_3$ ), 7.31 (mc, $8H_{ar}$ ), 7.90 (m, $H_{ar}$ -10) (c)
IIIc	244.5 (4.44), 272 (4.29), 307 sh (3.85), 345 (4.04), 390 (4.13)	1738 1622 1512 1732	3.32 (near s, $\text{CH}_25$ ), 7.20 (mc, $13\text{H}_{ar}$ ), 7.92 (m, $\text{H}_{ar}10$ ) (b)

<sup>(</sup>a) From ethyl acetate. (b) In tetrachloromethane. (c) In deuteriochloroform.

Compound No.	$NR_2$	Yield %	M.p., °C	Molecular Formula	Analyses % Calcd./Found		
					С	H	N
Va	$N[CH(CH_3)_2]_2$	40	144 (a)	$\mathrm{C_{14}H_{21}Cl_{2}NO_{2}S}$	49.71 49.53	$6.26 \\ 6.22$	4.14 4.00
Vb	$N(CH_3)C_6H_5$	36	156 (b)	$C_{15}H_{15}Cl_2NO_2S$	52.33 $52.17$	4.39 4.21	4.07 4.35
Ve	$N(C_6H_5)_2$	42	184 (a)	$C_{20}H_{17}Cl_2NO_2S$	59.12 59.11	4.21 4.48	3.44 3.40

#### Uv, Ir and Nmr Spectral Data

	Uv $\lambda$ max nm (log $\epsilon$ )	Ir, cm <sup>-1</sup> C=O C=C	Nmr, δ
Va	230 (3.52)	1792 1700 (c)	1.09 and 1.15 (2d, J = 6.6, 4CH <sub>3</sub> ), 2.80 (mc, $\rm CH_2\text{-}7 + CH_2\text{-}8 + 2NCH)$ , 3.28 (m, $\rm CH_2\text{-}5$ ), 3.63 (m, CH-4) (c)
Vb	244.5 (4.15), 292 (3.32)	1795 1708 (c)	2.70 (near s, NCH <sub>3</sub> ), 2.84 (mc, 3CH <sub>2</sub> ), 4.69 (m, CH-4), 6.91 (mc, $5H_{ar}$ ) (c)
Vc	250 (4.15), 293 sh (3.60)	1787 1712 (d)	2.27 and 2.72 (2m, $CH_2$ -7 + $CH_2$ -8), 3.22 and 3.55 (2d, J = 15, $CH_2$ -5), 5.08 (m, $CH$ -4), 7.08 and 7.25 (2m, $2NC_6H_5$ ) (e)

(a) From ethyl acetate. (b) From anhydrous diethyl ether. (c) In tetrachloromethane. (d) In chloroform. (e) In deuteriochloroform.

Table IV

N,N-Disubstituted 4-Amino-3-chloro-7,8-dihydro-2H,5H-thiopyrano[4,3-b] pyran-2-ones Vla-c

			S S CI					
			√0		Analys	es % Calcd./	Found	
Compound No.	NR <sub>2</sub>	Yield %	M.p., °C	Molecular Formula	С	H	N	
Vla	$N[CH(CH_3)_2]_2$	71	114 (a)	$\mathrm{C_{14}H_{20}CINO_{2}S}$	55.71 55.96	6.68 6.38	4.64 4.60	
VIb	$N(CH_3)C_6H_5$	85	151 (a)	$C_{15}H_{14}CINO_2S$	58.53 58.40	4.58 4.60	4.55 4.44	
VIc	$N(C_6II_5)_2$	51	150 (a)	$C_{20}H_{16}CINO_2S$	64.95 65.08	4.36 4.71	3.79 3.80	
		Uv,	Ir and Nmr Spectral	Data				
	Uv $\lambda$ max nm (log $\epsilon$ )	C	Ir, cm <sup>-1</sup> C=O C=C		Nmr, δ			
VIa	237.5 sh (3.30), 316 (3.79	) 1	735 1633 1510 (b)	1.21 (2 superimposed d, $J = 6.6$ , $4CH_3$ ), 2.85 (near s, $CH_2$ -7 + $CH_2$ -8), 3.46 (near s, $CH_2$ -5), 3.70 (h, $J = 6.6$ , 2NCH) (b)				
VIb	241.5 (4.11), 318 (3.87) 360 (3.79)	1	710 1632 1520 (c)	2.86 (m, $CH_2$ -7 + $CH_2$ -8), 3.08 (m, $CH_2$ -5), 3.33 (near s, $NCH_3$ ), 6.82 (mc, $NC_6H_5$ ) (d)				
VIc	253 (4.02), 279.5 (4.19), 320 (3.84), 374 (3.83)	I	710 1633 1517 (c)	2.84 (m, CH <sub>2</sub> -7 + C <sub>1</sub> 7.23 (2m, 2NC <sub>6</sub> H <sub>5</sub> )		n, CH <sub>2</sub> -5), 7.1	10 and	

(a) From ethyl acetate. (b) In tetrachloromethane. (c) In chloroform. (d) In deuteriochloroform.

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